The Mechanism of the Formation of Tropolone from the Adduct of Cyclopentadiene and Dichloroketene*1

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We have previously reported¹⁾ the novel rearrangements of the adduct of 6,6-dimethylfulvene with dichloroketene, and have proposed a mechanism for the formation of tropone derivatives from the adduct involving a norcaradienone intermediate. We have also proposed a similar mechanism for the rearrangement²⁾ of the adduct of cyclopentadiene with dichloroketene to tropolone. This report will present evidence to support the proposed mechanism by a tracer experiment.

The reaction of cyclopentadiene with dichloroketene-2- 14 C, produced by the reaction of the dichloroacetyl chloride- $^{2-14}$ C and triethylamine, in boiling *n*-hexane afforded a 1:1 adduct (1).^{2,3)}

The adduct (1) was rearranged to ¹⁴C containing tropolone (2) under solvolytic condition (potassium acetate in acetic acid and water) according to the method of Stevens *et al.*²⁾

The tropolone was submitted to catalytic hydrogenation in the presence of the Adams catalyst to give ¹⁴C containing cycloheptane-1,2-diol (3). The diol (3) was degraded to trimethylenediamine (7) and carbon dioxide via pimelic acid (4), pentamethylenediamine (5), and glutaric acid (6) by repeated oxidations and Schmidt reactions, as is shown in the following scheme.

Pimelic acid (4) shows a radioactivity*3 of 152,883 dpm/mM; the activities of the consecutive degradation products or their derivatives are shown in parentheses in terms of a percentage of the activity based on the activity of pimelic acid.

It is clear from the above results that all the ¹⁴C are located at the 3- or 7-carbon in the tropolone (2); this supports the previously-proposed mechanism for the formation of tropolone by way of the attack of the nucleophile at the 7-position of the norcaradienone intermediate (8). The details will be reported elsewhere.

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¹⁾ T. Asao, T. Machiguchi, T. Kitamura and Y. Kitahara, Chem. Commun., 1970, 89.

²⁾ H. C. Stevens, D. A. Reich, D. R. Brandt, K. R. Fountain and E. J. Gaughan, J. Amer. Chem. Soc., 87, 5257 (1965).

³⁾ L. Ghosez, R. Montaigne and P. Mollet, Tetrahedron Lett., 1966, 135.

^{*3} The activities of the compounds (4), (6) and dibenzoate of (5) and (7) were measured in dioxane, and those of BaCO₃ were measured in toluene on TEN Liquid Scintillation Counter GSL-111.